

1. A parallel plate capacitor has a charge of  $0.020 \mu\text{C}$  on each plate with a potential difference of  $240 \text{ V}$ . The parallel plates are separated by  $0.40 \text{ mm}$  of air. (a) What is the capacitance for this capacitor? (b) What is the area of a single plate? (c) What voltage will the air between the plates become ionized? Assume a dielectric strength of  $3.0 \text{ kV/mm}$  for air.

Solution.

- (a) Find the capacitance.

$$C = Q/V = (0.02 \times 10^{-6} \text{ C})/(240 \text{ V}) = 8.3 \times 10^{-11} \text{ F} = 83 \text{ pF}.$$

- (b) Find the area.

$$C = \epsilon_0 A/d;$$

$$(8.3 \times 10^{-11} \text{ F}) = (8.85 \times 10^{-12} \text{ C}^2/\text{N}\cdot\text{m}^2)A/(0.40 \times 10^{-3} \text{ m}),$$

$$A = 3.8 \times 10^{-3} \text{ m}^2$$

- (c) Find the voltage between the plates.

$$\Delta V = Ed = (3.0 \text{ kV/mm})(0.40 \text{ mm}) = 1.2 \text{ kV}.$$

2. A parallel plate capacitor is attached to a battery that supplies a constant voltage. While the battery is still attached, a dielectric of dielectric constant  $\kappa = 3.0$  is inserted so that it just fits between the plates. What is the energy stored in the capacitor after the dielectric is inserted in terms of the energy  $U_0$  before the dielectric was inserted?

Solution.

When a dielectric is inserted between the plates, the distance between the plates and the area of each plate remain unchanged, the capacitance of the capacitor increases to

$$C = \kappa C_0, \text{ where } C_0 \text{ is the capacitance of the capacitor before the dielectric is inserted.}$$

Assume the voltage supplied by the battery is  $V$ , and keeps unchanged.

The energy stored in the original capacitor is

$$U_0 = \frac{1}{2} C_0 V^2.$$

After the dielectric is inserted, the energy stored in the capacitor is

$$U = \frac{1}{2} C V^2 = \frac{1}{2} \kappa C_0 V^2 = \kappa U_0 = 3.0 U_0$$

3. (a) Calculate the capacitance per unit length of an axon of radius  $5.0 \mu\text{m}$ . The membrane acts as an insulator between the conducting fluids inside and outside the neuron. The membrane is  $6.0 \text{ nm}$  thick and has a dielectric constant of  $7.0$ . (Note: The membrane is thin compared to the radius of the axon, so the axon can be treated as a parallel plate capacitor.) (b) In its resting state (no signal being transmitted), the potential of the fluid inside is about  $85 \text{ mV}$  lower than the outside.

Therefore, there must be small net charges  $\pm Q$  on either side of the membrane. Which side has positive charge? What is the magnitude of the charge density on the surfaces of the membrane.

Solution.

- (a) The area of the curved surface of a cylinder (axon) is  $2\pi rL$  where  $r$  and  $L$  are the radius and length of a cylinder, respectively. Then find the capacitance of the cylinder:

$$C = \kappa \epsilon_0 A/d = [\kappa \epsilon_0 (2\pi rL)]/d$$

$$C/L = 2\pi \kappa \epsilon_0 r/d = 2\pi (7.0)(8.85 \times 10^{-12} \text{ C}^2/\text{N}\cdot\text{m}^2)(5.0 \times 10^{-6} \text{ m})/(6.0 \times 10^{-9} \text{ m})$$

$$= 3.2 \times 10^{-7} \text{ F/m}.$$

- (b) The outside of the membrane has the positive charge, since the potential is higher outside than inside.

$$\Delta V = Ed$$

$$= (\sigma/\kappa \epsilon_0)d$$

$$\text{Then } \sigma = (\kappa \epsilon_0 \Delta V)/d$$

$$= (7.0)(8.85 \times 10^{-12} \text{ C}^2/\text{N}\cdot\text{m}^2)(0.085 \text{ V})/(6.0 \times 10^{-9} \text{ m})$$

$$= 8.8 \times 10^{-4} \text{ C/m}^2$$

4. Two electrodes are placed in a calcium chloride solution and a potential difference is maintained between them. If  $3.8 \times 10^{16} \text{ Ca}^{2+}$  ions and  $6.2 \times 10^{16} \text{ Cl}^-$  ions per second move in opposite directions through an imaginary area between the electrodes, what is the current in the solution? Solution.

Since the oppositely-charged ions move in opposite directions, they both contribute to the current in the same direction.

$$I = \Delta q/\Delta t$$

$$= Ne/\Delta t$$

$$= [2(3.8 \times 10^{16}) + 6.2 \times 10^{16}](1.602 \times 10^{-19} \text{ C/s})$$

$$= 22.1 \text{ mA}.$$

5. A 1.5-V battery provides 0.5 A of current. How much chemical energy is consumed during 5 minutes?

Solution.

The work done by the battery is

$$\begin{aligned}W &= \mathcal{E}q \\ &= \mathcal{E}It \\ &= (1.5 \text{ V})(0.5 \text{ A})(5 \text{ min})(60 \text{ s/min}) \\ &= 225 \text{ J} = -\Delta U_{\text{chemical}}\end{aligned}$$