

answers

- IC 1. sodium acetate $NaC_2H_3O_2$ MC/BI 2. carbon monoxide CO
- MC/A 3. hydriodic acid $HI (ag)$ MC/A 4. hydrogen iodide $HI(g)$
- E 5. bromine Br_2 IC 6. chromium (III) nitrate $Cr(NO_3)_3$
- IC 7. zinc hydroxide $Zn(OH)_2$ MC/BI 8. phosphorus pentachloride PCl_5
- IC 9. aluminum hydrogen carbonate $Al(HCO_3)_2$ E 10. sodium Na
- MC/A 11. carbonic acid H_2CO_3 IC 12. ammonium phosphate $(NH_4)_3PO_4$
- E 13. iron Fe IC 14. iron (III) carbonate $Fe_2(CO_3)_3$
- MC/BI 15. silicon tetrachloride $SiCl_4$ MC/A 16. nitric acid HNO_3
- MC/A 17. hydrogen sulfide $H_2S (g)$ MC/A 18. hydrosulfuric acid $H_2S (ag)$
- MC/A 19. sulfurous acid $H_2SO_3 (g)$ MC/A 20. sulfuric acid H_2SO_4
- IC 21. manganese (II) sulfide MnS IC 22. ammonium sulfate $(NH_4)_2SO_4$
- IC 23. mercury (II) hydroxide $Hg(OH)_2$ MC/BI 24. sulfur dioxide SO_2
- E 25. hydrogen H_2 MC/BI 26. dihydrogen monoxide H_2O
- IC 27. aluminum acetate $Al(C_2H_3O_2)_3$ MC/BI 28. sulfur trioxide SO_3
- E 29. mercury Hg IC 30. copper (II) nitrate $Cu(NO_3)_2$
- MC/A 31. acetic acid $HC_2H_3O_2$ MC/BI 32. carbon tetrachloride CCl_4
- IC 33. zinc hydrogen carbonate $Zn(HCO_3)_2$ IC 34. calcium perchlorate $Ca(ClO_4)_2$
- IC 35. tin (IV) nitride Sn_3N_4 E 36. calcium Ca
- IC 37. calcium phosphate $Ca_3(PO_4)_2$ MC/A 38. hydrofluoric acid $HF (ag)$
- MC/BI 39. oxygen difluoride OF_2 IC 40. iron (III) hydroxide $Fe(OH)_3$