

Determine the type of substance and then name the substance IC = ionic compound MC/A = molecular compound/acid E = element MC/BI = molecular compound/binary inorganic

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|---|---|
| 1. NaCl | 2. CO ₂ |
| 3. HNO ₃ | 4. Cu |
| 5. Cl ₂ | 6. Na ₃ PO ₄ |
| 7. H ₂ SO ₃ | 8. (NH ₄) ₂ SO ₄ |
| 9. N ₂ O ₅ | 10. P ₄ |
| 11. H ₃ PO ₄ | 12. Ca ₃ (PO ₄) ₂ |
| 13. P ₄ O ₁₀ | 14. FeBr ₃ |
| 15. FeBr ₂ | 16. AgNO ₃ |
| 17. SF ₆ | 18. S ₈ |
| 19. H ₂ CO ₃ | 20. NaHCO ₃ |
| 21. HCl (aq) | 22. HCl (g) |
| 23. Al (OH) ₃ | 24. N ₂ O |
| 25. H ₂ SO ₄ | 26. H ₂ SO ₄ |
| 27. BaI ₂ | 28. LiOH |
| 29. NiO | 30. N ₂ O ₃ |
| 31. PbS ₂ | 32. HC ₂ H ₃ O ₂ |
| 33. SnSO ₃ | 34. K ₂ CO ₃ |
| 35. PCl ₃ | 36. HNO ₃ |
| 37. HBr (aq) | 38. HBr (g) |
| 39. Fe ₂ (SO ₄) ₃ | 40. FeSO ₄ |
| 41. NaClO ₄ | 42. HClO ₄ |